

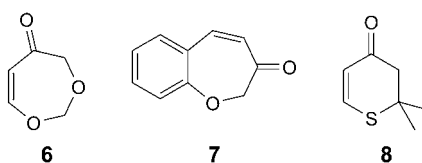
Photocyclodimers of ‘*Made-to-Measure*’ Seven- and Six-Membered Cyclic Enones

by Kerstin Schmidt and Paul Margaretha*

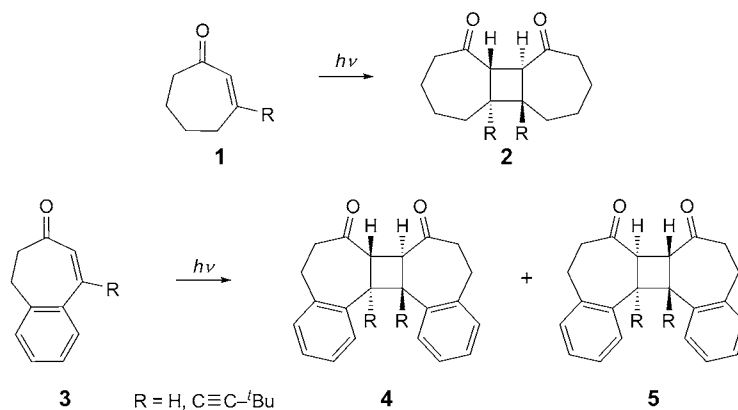
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On irradiation (350 nm) in benzene as solvent, dioxepinone **6** and benzoxepinone **7** afford quantitatively mixtures of two diastereoisomeric *head-to-head* dimers, respectively. In both cases, on contact with SiO₂ the minor dimer containing *trans*-ring fusions undergoes spontaneous isomerization to the (major) *cis-transoid-cis* diastereoisomer. In contrast, thiopyranone **8** is converted selectively, but in very low yield, to dimer **13**.

Introduction. – Synthetic applications and mechanistic details of both photocyclodimerization and photocycloaddition to alkenes have been thoroughly investigated for five- and six-membered cyclic enones [1–3]. In contrast, information on the corresponding behavior of cyclohept-2-enones is restricted to the finding that such seven-membered cyclic enones **1** on irradiation afford predominantly doubly *trans*-fused dimers **2**, the formation of which has been proposed to occur *via* (*Z/E*)-isomerization of the excited enone, followed by thermal [$2\pi_s + 2\pi_a$] cycloaddition to a second ground-state enone [4–6]. Benzocyclohepten-7-ones **3** behave similarly to give diastereoisomer mixtures of dimers **4** and **5** [6][7] (*Scheme 1*). We have recently reported that dioxepinone **6** and benzoxepinone **7** undergo [2 + 2] photocycloaddition to 1,3-dienes, *e.g.*, 2,3-dimethylbuta-1,3-diene [8] without any or very little interference of photodimer formation. Whereas photocycloaddition to 1,3-dienes is atypical for cyclopentenones and cyclohexenones due to very efficient energy transfer from the excited enone to the diene, more flexible six-membered ring enones, *e.g.*, thiopyranone **8**, have been found to give cyclobutanes almost quantitatively under these conditions [9]. For all these heterocyclic enones, the *a*) a shorter C–O bond, or *b*) a longer C–S bond as compared to a standard C–C bond seems to control the rigidity of the ring to prevent the above mentioned typical (*Z/E*)-isomerization of cycloheptenones. Here, we report on the photocyclodimerization of these oxa- and thia-enones.



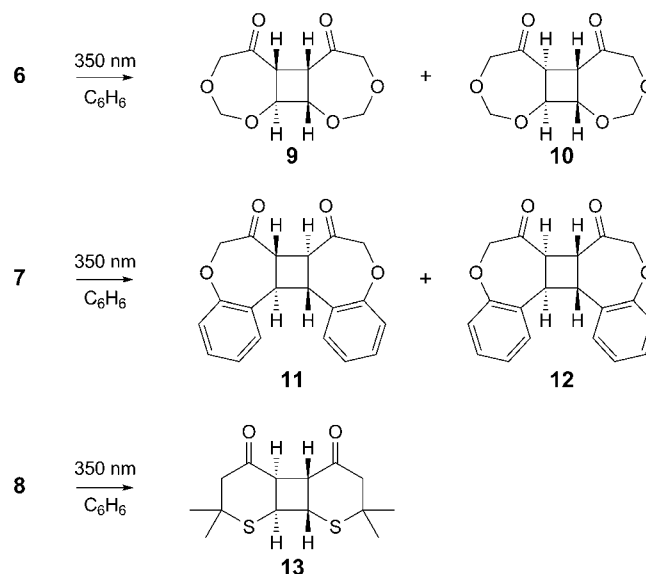
Scheme 1



Results. – Irradiation (350 nm) of a 1M solution of 1,3-dioxepin-5(4*H*)-one (**6**) in benzene for 24 h afforded a 2 : 3 mixture of dimers **9** and **10**, as monitored by ¹H-NMR. On standing in CDCl₃ or on treatment with SiO₂ in CH₂Cl₂, dimer **9** underwent isomerization quantitatively to **10**. Similarly, irradiation of a 1M solution of 1-benzoxepin-3(2*H*)-one (**7**) for 24 h led to a 2 : 3 mixture of dimers **11** and **12**; the minor product **11** again isomerized quantitatively to **12** in the presence of acids. The structures of dimers **10** and **12** were established by X-ray crystal-structure determination. In contrast, the photodimerization of 2,3-dihydro-2,2-dimethyl-4*H*-thiopyran-4-one (**8**) required much longer irradiation times (80–90 h for *ca.* 20% conversion) to give dimer **13** selectively (Scheme 2).

Discussion. – By assuming that *trans-anti-trans*-dimers, *e.g.* **2** or **4**, were formed solely *via* the sequence *a*) photoisomerization of the (excited) (*Z*)-cycloheptenone to the (ground state) (*E*)-diastereoisomer, followed by *b*) thermal [₂π_s + ₂π_a] cycloaddition to a second ground-state enone, the behavior of benzoxepinone **7** parallels that of benzocycloheptenones **3**. The effect of the shorter C–O–C bonds in **11** as compared to the carbocyclic dimer **4** is reflected in the spontaneous isomerization of the former to **12** in contact with SiO₂, while the latter is apparently stable under these conditions. In contrast, a comparison of the results on the photodimerization of dioxepinone **6** with those of cycloheptenones **1** indicates, that (*Z/E*)-isomerization of the excited dioxanone was inhibited, as no doubly *trans*-fused tricyclic dimer was formed at all. Indeed, the behavior of **6** rather corresponds to that of six-membered cyclic oxa-enones, *e.g.*, 2,3-dihydro-2,2-dimethyl-4*H*-pyran-4-one [10]. Here, the *twisted* excited enone is trapped by a second ground-state enone molecule to yield a mixture of a *trans-anti-cis*-dimer **9**, wherein (only) one of the external rings is *trans*-fused to the (central) cyclobutane ring, in addition to the (conventional) *cis-anti-cis*-dimer **10**. In this context, the inefficiency observed in the photodimerization of thiopyranone **8** is difficult to rationalize. Thus, in contrast to their carbocyclic counterparts, 1,3-dioxepin-5(4*H*)-ones can be regarded as ‘*made-to-measure*’ seven-membered ring enones. Due to the shorter

Scheme 2



C–O bond length (vs. C–C), the rigidity of the ring system is increased, thereby preventing the (typical) (*Z/E*)-photoisomerization of cyclohept-2-enones.

Experimental Part

1. *General*. Photolyses were performed in a *Rayonet RPR-100* photoreactor equipped with (16) 350-nm lamps using solvents of spectrophotometric grade. Column chromatography (CC) was performed on silica gel (*Merck* grade 60; 230–400 mesh). ¹H- and ¹³C-NMR spectra (including 2D plots): *Bruker WM-500* at 500.13 and 125.8 MHz, resp.; δ in ppm rel. to Me₄Si as internal standard, *J* in Hz. X-Ray analyses: *Bruker SMART APEX II* three-circle diffractometer at 153 K with MoK α radiation (λ 0.71073 Å).

2. *Starting Materials*. Compounds **6** [11], **7** [12], and **8** [13] were synthesized according to the literature procedures.

3. *Photocyclodimerizations*. Ar degassed 5 · 10⁻¹ M solns. of **6**, **7**, or **8** in benzene were irradiated for the time indicated up to the degree of conversion determined by ¹H-NMR monitoring. Workup of the crude product mixtures as described below.

Irradiation of 6. A soln. of 114 mg of **6** in benzene (2 ml) was irradiated for 18 h up to total conversion. After filtering from polymeric material, subsequent evaporation of the solvent afforded 80 mg of a 2:3 mixture of dimers **9** and **10** (¹H-NMR).

rac-(5aR,5bS,10aS,10bS)-Tetrahydrocyclobuta[1,2-d:4,3-d']bis[1,3]dioxepine-5,6(4H,7H)-dione (**9**). ¹H-NMR (CDCl₃; from the mixture **9/10**): 5.50, 4.68 (AB, *J* = 7.3, CH₂(9)); 5.07, 4.93 (AB, *J* = 5.3, CH₂(2)); 4.65 (dd, *J* = 4.1, 6.5, H–C(5b)); 4.54, 4.20 (AB, *J* = 18.3, CH₂(7)); 4.44, 4.08 (AB, *J* = 18.2, CH₂(4)); 4.33 (dd, *J* = 9.3, 9.8, H–C(10b)); 3.98 (dd, *J* = 6.5, 9.8, H–C(10a)); 3.37 (dd, *J* = 4.4, 9.3, H–C(5a)). ¹³C-NMR (CDCl₃): 208.7 (s, C(5)); 206.4 (s, C(6)); 101.5 (t, C(9)); 97.1 (t, C(2)); 80.0 (t, C(7)); 75.1 (t, C(4)); 74.1 (d, C(5b)); 73.5 (d, C(10a)); 59.0 (d, C(10b)); 45.2 (d, C(5a)).

The crude residue was dissolved in CH₂Cl₂ (1 ml), added to a suspension of 50 mg of SiO₂ in CH₂Cl₂ (2 ml), and the mixture was stirred for 3 h at r.t. After filtering and evaporation of the solvent, recrystallization from hexane afforded 55 mg (48%) of **10**.

rac-(5aS,5bS,10aS,10bS)-Tetrahydrocyclobuta[1,2-d:4,3-d']bis[1,3]dioxepine-5,6(4H,7H)-dione (**10**). M.p. 208–210°. ¹H-NMR (CDCl₃): 4.97, 4.80 (AB, *J* = 5.3, CH₂(2), CH₂(9)); 4.34, 3.98 (AB, *J* = 17.8, CH₂(4), CH₂(6)); 4.32, 4.13 (AA'XX', *J*_{AX} = 8.1, *J*_{AA'} = 5.7, *J*_{XX'} = 3.0, *J*_{AX'} = 0.5, H–C(5a), H–C(5b), H–C(10a), H–C(10b)). ¹³C-NMR (CDCl₃): 209.6 (s, C(5), C(6)); 96.1 (t, C(2), C(9)); 76.1 (t, C(4), C(7)); 74.2 (d, C(5a), C(5b)); 47.5 (d, C(10a), C(10b)).

*X-Ray Crystal-Structure Determination of 10*¹⁾. C₁₀H₁₂O₆, orthorhombic, space group *P2₁2₁1*, *a* = 6.5219(2), *b* = 8.5138 (2), *c* = 17.3353(5) Å.

Irradiation of 7. A soln. of 160 mg of **7** in benzene (2 ml) was irradiated for 18 h up to total conversion. After filtering from polymeric material, subsequent evaporation of the solvent afforded 120 mg of a 2:3 mixture of dimers **11** and **12** (¹H-NMR).

rac-(7aS,7bS,14bR,14cR)-7a,7b,14b,14c-Tetrahydrodibenzo[b,b']cyclobuta[1,4-d:2,3-d']bisoxepine-7,8(6H,9H)-dione (**11**). ¹H-NMR (CDCl₃; from the mixture **11/12**): 7.35–7.13 (8 arom. H); 4.68, 4.46 (AB, *J* = 17.1, CH₂(6), CH₂(9)); 3.88, 3.80 (2*m*, H–C(7a), H–C(7b), H–C(14b), H–C(14c)). ¹³C-NMR (CDCl₃): 206.2 (s, C(7), C(8)); 158.9 (s, C(4a), C(10a)); 133.5 (s, C(14a), C(14d)); 129.2, 128.8, 127.5, 126.9 (4*d*, 8 arom. CH); 49.2 (d, C(7a), C(7b)); 43.1 (d, C(14b), C(14c)).

The crude residue was dissolved in CH₂Cl₂ (1 ml), added to a suspension of 50 mg SiO₂ in CH₂Cl₂ (2 ml), and the mixture was stirred for 3 h at r.t. After filtering and evaporation of the solvent, recrystallization from hexane/benzene 2:1 afforded **12** (85 mg, 52%).

rac-(7aR,7bR,14bR,14cR)-7a,7b,14b,14c-Tetrahydrodibenzo[b,b']cyclobuta[1,4-d:2,3-d']bisoxepine-7,8(6H,9H)-dione (**12**). M.p. 245–247°. ¹H-NMR (CDCl₃): 7.40–7.15 (8 arom. H); 4.74, 4.28 (AB, *J* = 16.4, CH₂(6), CH₂(9)); 4.58, 4.38 (AA'XX', *J*_{AX} = 11, *J*_{AA'} = 9, *J*_{XX'} = 4, *J*_{AX'} = 1, H–C(7a), H–C(7b), H–C(14b), H–C(14c)). ¹³C-NMR (CDCl₃): 209.1 (s, C(7), C(8)); 158.2 (s, C(4a), C(10a)); 133.4 (s, C(14a), C(14d)); 129.2, 128.8, 127.5, 126.9 (4*d*, 8 arom. CH); 46.0 (d, C(14b), C(14c)); 42.2 (d, C(7a), C(7b)).

*X-Ray Crystal-Structure Determination of 12*¹⁾. C₂₀H₁₆O₄, orthorhombic, space group *P2₁2₁2₁*, *a* = 5.216(2), *b* = 10.846(4), *c* = 26.570(10) Å.

Irradiation of 8. A soln. of 142 mg of **8** in benzene (2 ml) was irradiated for 90 h. After filtering from polymeric material, evaporation of the solvent afforded a 4:1 mixture of **8** and dimer **13** (¹H-NMR). CC (SiO₂; pentane/Et₂O 3:2) afforded first **8**, followed by **13** (15 mg, 10%).

rac-(4aR,4bR,8aS,8bS)-2,2,7,7-Tetramethyloctahydrocyclobuta[1,2-b:4,3-b']bisthiopyran-4,5-dione (**13**). M.p. 120–121°. ¹H-NMR (CDCl₃): 3.95, 3.72 (AA'XX', *J*_{AX} = 8.6, *J*_{AA'} = 6.9, *J*_{XX'} = 5.2, *J*_{AX'} = 0.4, H–C(4a), H–C(4b), H–C(8a), H–C(8b)); 2.67, 2.63 (AB, *J* = 14.3, CH₂(3), CH₂(6)); 1.41, 1.33 (2*s*, 4 Me). ¹³C-NMR (CDCl₃): 208.1 (s, C(4), C(5)); 56.2 (t, C(3), C(6)); 48.3 (d, C(4a), C(4b)); 47.2 (d, C(8a), C(8b)); 44.8 (s, C(2), C(7)); 31.1, 30.2 (2*q*, 4 Me).

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¹⁾ CCDC-840649 and -840650 contain the supplementary crystallographic data for **12** and **10**, respectively. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif.

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